

## THE PERIODIC TABLE OF ELEMENTS

## Group

																		0																	
1		2																2																	
7	<b>Li</b> Lithium	9	<b>Be</b> Beryllium													4	<b>He</b> Helium	2																	
3		4														5		10																	
23	<b>Na</b> Sodium	24	<b>Mg</b> Magnesium													14	<b>Si</b> Silicon	18																	
11		12														15		36																	
39	<b>K</b> Potassium	40	<b>Ca</b> Calcium	45	<b>Sc</b> Scandium	48	<b>Ti</b> Titanium	51	<b>V</b> Vanadium	52	<b>Cr</b> Chromium	55	<b>Mn</b> Manganese	56	<b>Fe</b> Iron	59	<b>Co</b> Cobalt	64	<b>Ni</b> Nickel	65	<b>Cu</b> Copper	70	<b>Ga</b> Gallium	73	<b>Ge</b> Germanium	75	<b>As</b> Arsenic	79	<b>Se</b> Selenium	80	<b>Br</b> Bromine	84	<b>Kr</b> Krypton		
19		20																																	
85	<b>Rb</b> Rubidium	88	<b>Sr</b> Strontium	89	<b>Y</b> Yttrium	91	<b>Zr</b> Zirconium	93	<b>Nb</b> Niobium	96	<b>Mo</b> Molybdenum	98	<b>Tc</b> Technetium	101	<b>Ru</b> Ruthenium	103	<b>Rh</b> Rhodium	106	<b>Pd</b> Palladium	108	<b>Ag</b> Silver	112	<b>Cd</b> Cadmium	115	<b>In</b> Indium	119	<b>Sn</b> Tin	122	<b>Sb</b> Antimony	128	<b>Te</b> Tellurium	127	<b>I</b> Iodine	131	<b>Xe</b> Xenon
37		38																																	
133	<b>Cs</b> Caesium	137	<b>Ba</b> Barium	139	<b>La</b> <sup>*</sup> Lanthanum	178	<b>Hf</b> Hafnium	181	<b>Ta</b> Tantalum	184	<b>W</b> Tungsten	186	<b>Re</b> Rhenium	190	<b>Os</b> Osmium	192	<b>Ir</b> Iridium	195	<b>Pt</b> Platinum	197	<b>Au</b> Gold	201	<b>Hg</b> Mercury	204	<b>Tl</b> Thallium	207	<b>Pb</b> Lead	209	<b>Bi</b> Bismuth	210	<b>Po</b> Polonium	210	<b>At</b> Astatine	222	<b>Rn</b> Radon
55		56																																	
223	<b>Fr</b> Francium	226	<b>Ra</b> Radium	227	<b>Ac</b> <sup>†</sup> Actinium	261	<b>Rf</b> Rutherfordium	262	<b>Db</b> Dubnium	266	<b>Sg</b> Seaborgium	264	<b>Bh</b> Bohrium	277	<b>Hs</b> Hassium	268	<b>Mt</b> Meitnerium	271	<b>Ds</b> Darmstadtium	272	<b>Rg</b> Roentgenium	285	<b>Cn</b> Copernicium												
87		88																																	
140	<b>Ce</b> Cerium	141	<b>Pr</b> Praseodymium	144	<b>Nd</b> Neodymium	145	<b>Pm</b> Promethium	150	<b>Sm</b> Samarium	152	<b>Eu</b> Europium	157	<b>Gd</b> Gadolinium	159	<b>Tb</b> Terbium	162	<b>Dy</b> Dysprosium	165	<b>Ho</b> Holmium	167	<b>Er</b> Erbium	169	<b>Tm</b> Thulium	173	<b>Yb</b> Ytterbium	175	<b>Lu</b> Lutetium								
58		59		60		61		62		63		64		65		66		67		68		69		70		71									
232	<b>Th</b> Thorium	231	<b>Pa</b> Protactinium	238	<b>U</b> Uranium	237	<b>Np</b> Neptunium	242	<b>Pu</b> Plutonium	243	<b>Am</b> Americium	247	<b>Cm</b> Curium	245	<b>Bk</b> Berkelium	251	<b>Cf</b> Californium	254	<b>Es</b> Einsteinium	253	<b>Fm</b> Fermium	256	<b>Md</b> Mendelevium	254	<b>No</b> Nobelium	257	<b>Lr</b> Lawrencium								
90		91		92		93		94		95		96		97		98		99		100		101		102		103									

\* 58 – 71 Lanthanum series  
† 90 – 103 Actinium series

a

x

b

a = relative atomic mass (approx)  
x = atomic symbol  
b = atomic number